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113 - Worksheet - Photons

**Diploma Questions – Photoelectric Effect (Basic Concepts)**

**Q673:** The wave theory of light can account for all of the following EXCEPT

- a. Reflection
- b. Refraction
- c. Polarization
- d. Photoelectric effect

**Q679:** A metal surface has a threshold frequency of  $f_0$ . If the metal surface is illuminated with radiation of a frequency of  $3.5f_0$ , then the maximum kinetic energy of emitted photoelectrons is

- a.  $4.5hf_0$
- b.  $3.5hf_0$
- c.  $2.5hf_0$
- d.  $1.0hf_0$

$$\begin{aligned}
 E_{\text{photon}} &\rightarrow W + E_K \\
 3.5hf_0 &\rightarrow hf_0 + E_K \\
 -hf_0 &\quad -hf_0 \\
 2.5hf_0 &= E_K
 \end{aligned}$$

**Q680:** If a photoelectric device is illuminated by light with a frequency greater than threshold frequency, what will happen as the frequency of the light is increased?

- a. The photoelectric current will increase.
- b. More photoelectrons will be emitted.
- c. The work function of the device will increase.
- d. Photoelectrons will be emitted with greater energy.

$$\begin{aligned}
 E_{\text{photon}} &\rightarrow W + E_K \\
 hf &\rightarrow W + E_K
 \end{aligned}$$

Single photon still only causes emission of a single photoelectron.

**Q682:** In a photoelectric experiment, the maximum kinetic energy of photoelectrons does **not** depend on the

- a. Work function of the emitted material
- b. Wavelength of the incident light
- c. Intensity of the incident light
- d. Energy of an incident photon

$$E_{\text{photon}} \rightarrow W + E_K$$

Diploma Questions – Photoelectric Effect (Calculations)

**Q685:** Sodium used in a photoelectric cell is irradiated by EM radiation of frequency  $6.2 \times 10^{15}$  Hz. If the threshold frequency for sodium is  $5.6 \times 10^{14}$  Hz, then the maximum  $E_k$  of the photoelectrons produced is

- a.  $1.1 \times 10^{-19}$  J
- b.  $4.1 \times 10^{-18}$  J
- c.**  $3.7 \times 10^{-18}$  J
- d.  $5.6 \times 10^{-15}$  J

$$E_{\text{photon}} \rightarrow W + E_k$$

$$hf \rightarrow hf_0 + E_k$$

$$(6.63 \times 10^{-34})(6.2 \times 10^{15}) = (6.63 \times 10^{-34})(5.6 \times 10^{14}) + E_k$$

$$E_k = 3.73932 \times 10^{-18} \text{ J}$$

**Q687:** If light with a frequency of  $1.0 \times 10^{15}$  Hz causes a metal to emit electrons with kinetic energy of  $1.0 \times 10^{-19}$  J, the work function of the metal is

- a.  $1.0 \times 10^{-4}$  J
- b.  $6.6 \times 10^{-19}$  J
- c.**  $5.6 \times 10^{-19}$  J
- d.  $1.0 \times 10^{-19}$  J

$$E_{\text{photon}} \rightarrow W + E_k$$

$$hf \rightarrow W + E_k$$

$$(6.63 \times 10^{-34})(1.0 \times 10^{15}) = W + (1.0 \times 10^{-19})$$

$$W = 5.63 \times 10^{-19} \text{ J}$$

**Q697:** Copper has a work function of 4.46 eV. What is the maximum kinetic energy of the ejected electrons if the metal is illuminated by light with a wavelength of 450 nm?

- a.  $2.72 \times 10^{-19}$  J
- b.  $4.42 \times 10^{-19}$  J
- c.  $7.14 \times 10^{-19}$  J
- d.** 0 J, because no electrons are ejected

$$E_{\text{photon}} \rightarrow W + E_k$$

$$\frac{hc}{\lambda} \rightarrow W + E_k$$

$$\frac{(6.63 \times 10^{-34})(3.0 \times 10^8)}{(450 \times 10^{-9})} = 7.136 \times 10^{-19} + E_k$$

$$4.42 \times 10^{-19} = 7.136 \times 10^{-19} + E_k$$



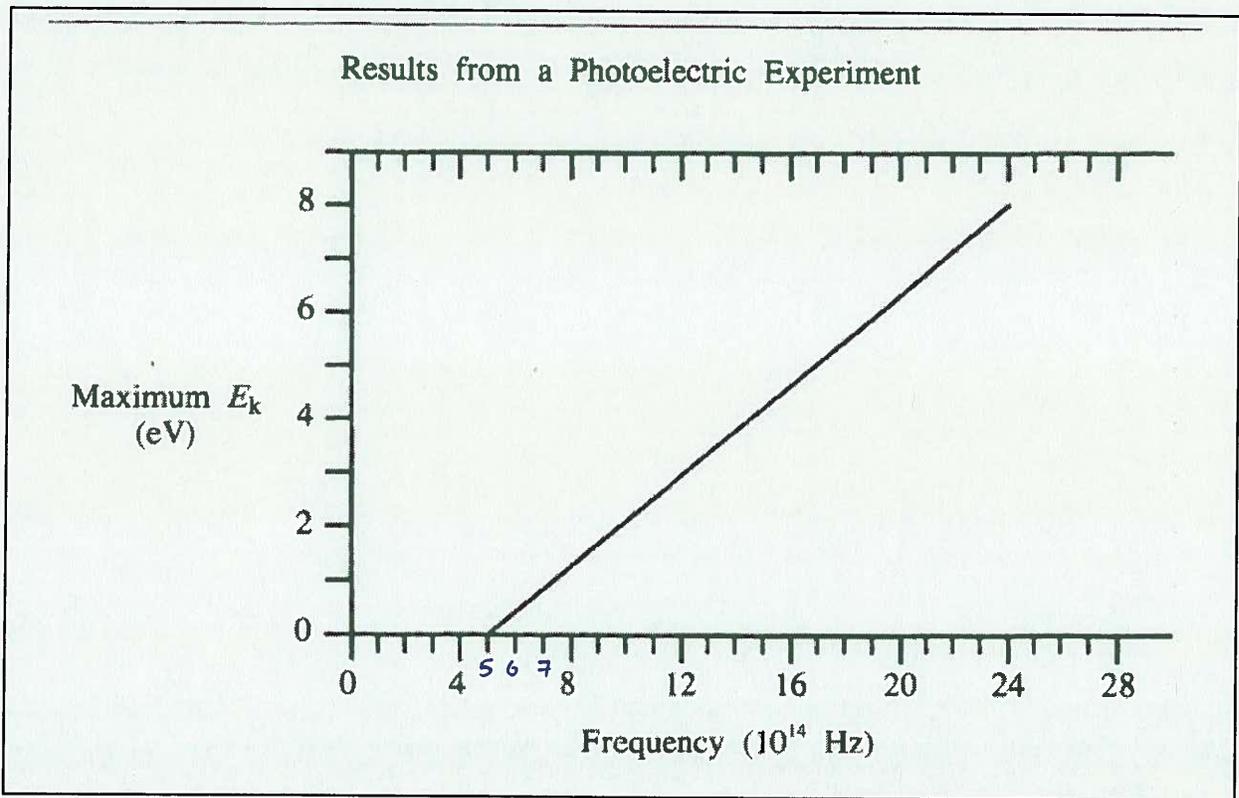
**Q710:** Violet light striking the negative electrode in a phototube causes a current to flow in the tube. Under the same conditions, another form of light that will always cause a current to flow is

- a. Blue
- b. Green
- c. Infrared
- d. Ultraviolet

Low f High f  
R M I V U X G

**Diploma Questions – Photoelectric Effect (Graphs)**

Use the following information to answer Q711:



**Q711:** The maximum kinetic energy of the photoelectrons that are emitted by light of wavelength  $1.7 \times 10^{-7}$  m is

- a. 1.1 eV
- b. 5.2 eV
- c. 6.2 eV
- d. 7.3 eV

$$f_0 = 5 \times 10^{14} \text{ Hz}$$

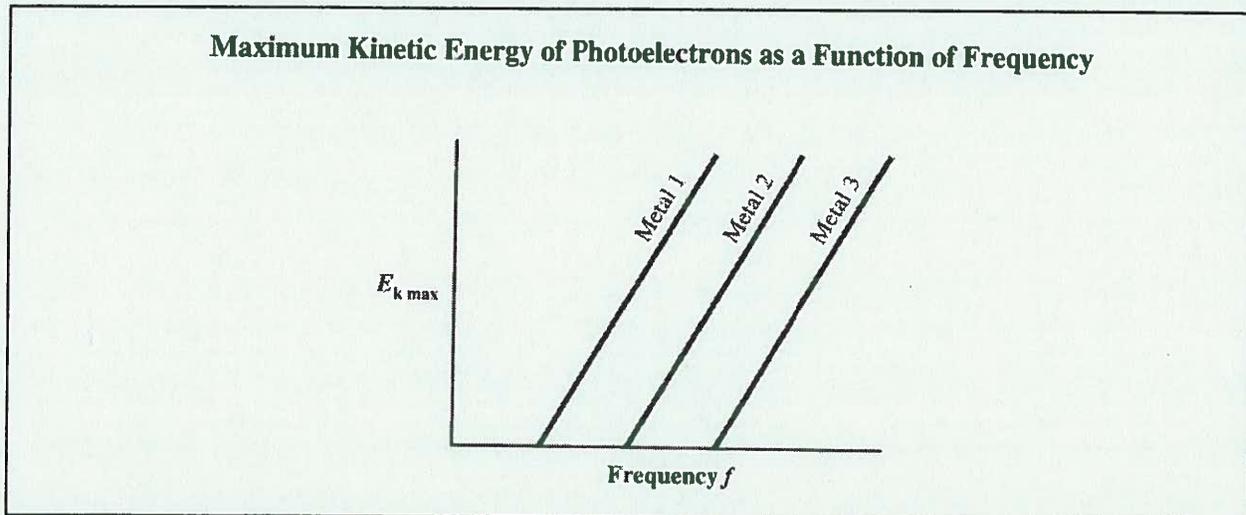
$$E_{\text{photon}} \rightarrow W + E_K$$

$$\frac{hc}{\lambda} \rightarrow hf_0 + E_K$$

$$\frac{(4.14 \times 10^{-15})(3.0 \times 10^8)}{(1.7 \times 10^{-7})} = (4.14 \times 10^{-15})(5 \times 10^{14}) + E_K$$

$$E_K = 5.23588... \text{ eV}$$

Use the following information to answer Q712:

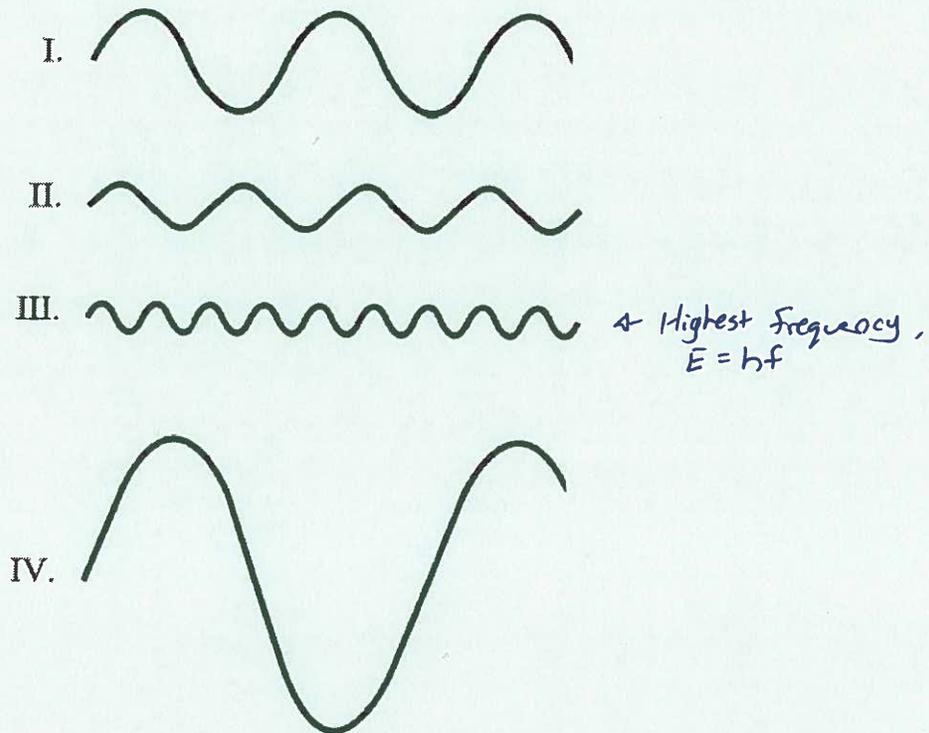


**Q712:** The graph illustrates that

- × a. For a given frequency of incident light, photoelectrons emitted from metal 2 have more kinetic energy than those emitted from metal 1
- × b. The maximum kinetic energy of emitted photoelectrons is dependent upon the intensity of the incident light
- × c. Metals will emit photoelectrons only if the intensity of the incident light is greater than some critical value
- Ⓓ Metals will emit photoelectrons only if the frequency of incident light is greater than some critical value

Use the following information to answer Q713:

These diagrams represent four electromagnetic waves incident upon the same metal surface.



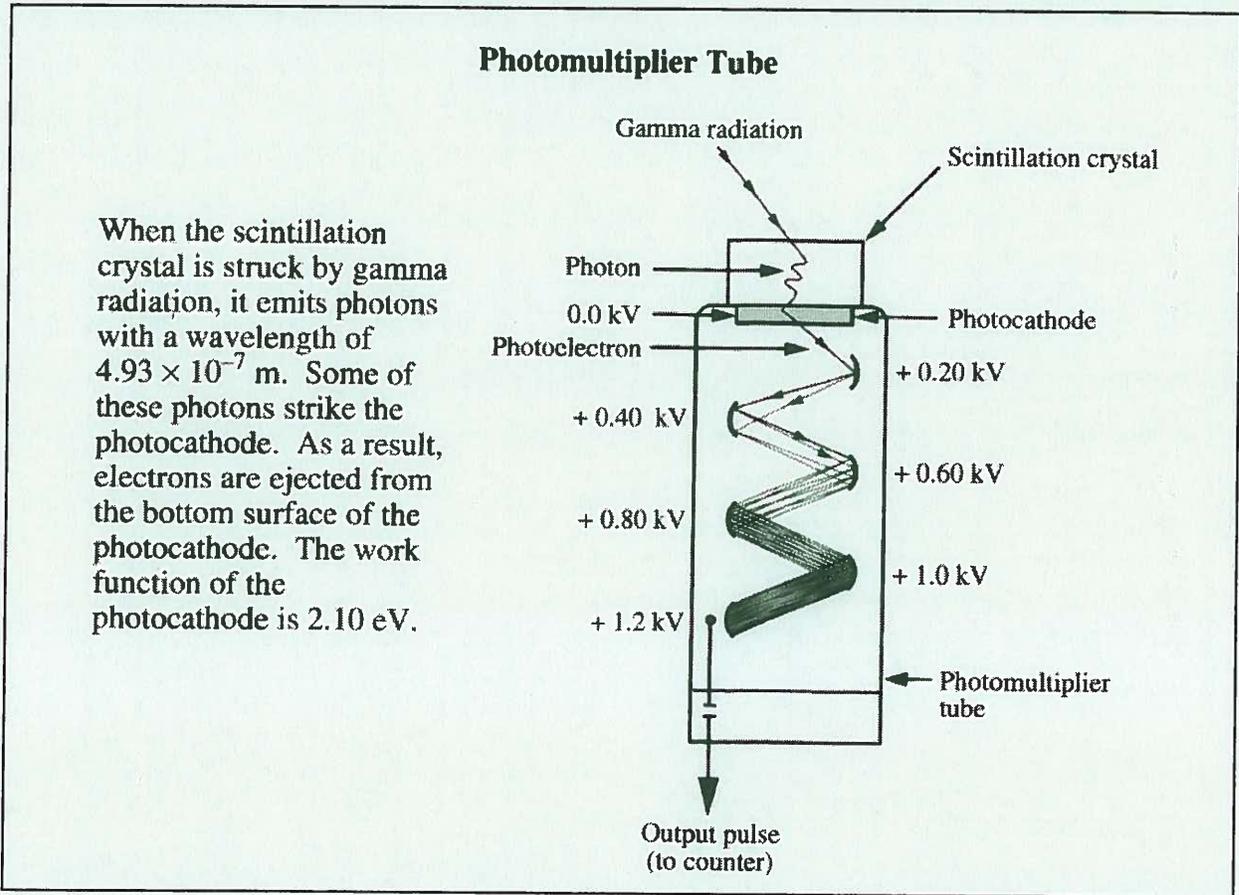
Amplitude is plotted vertically and time is plotted horizontally. All diagrams are drawn to the same scale.

**Q713:** Photoelectrons with the greatest kinetic energy would be emitted as the result of wave

- a. I
- b. II
- c. III
- d. IV

**Diploma Questions – Photoelectric Effect (in Technology)**

Use the following information to answer Q725 – Q726:



**Q725:** The maximum kinetic energy of the electrons ejected from the photocathode is

- a.  $6.7 \times 10^{-20}$  J
- b.  $3.4 \times 10^{-19}$  J
- c.  $4.0 \times 10^{-19}$  J
- d.  $7.4 \times 10^{-19}$  J

$$E_{\text{photon}} \rightarrow W + E_K$$

$$\frac{hc}{\lambda} \rightarrow W + E_K$$

$$\frac{(6.63 \times 10^{-34})(3.0 \times 10^8)}{(4.93 \times 10^{-7})} = 3.36 \times 10^{-19} + E_K$$

$$E_K = 6.7448275862 \times 10^{-20} \text{ J}$$

**Q726:** The electrons leaving the photocathode are attracted by the 0.20 kV electrode. The maximum speed they attain is

- a.  $8.6 \times 10^5$  m/s
- b.  $8.4 \times 10^6$  m/s
- c.  $7.0 \times 10^{13}$  m/s
- d.  $2.1 \times 10^{15}$  m/s

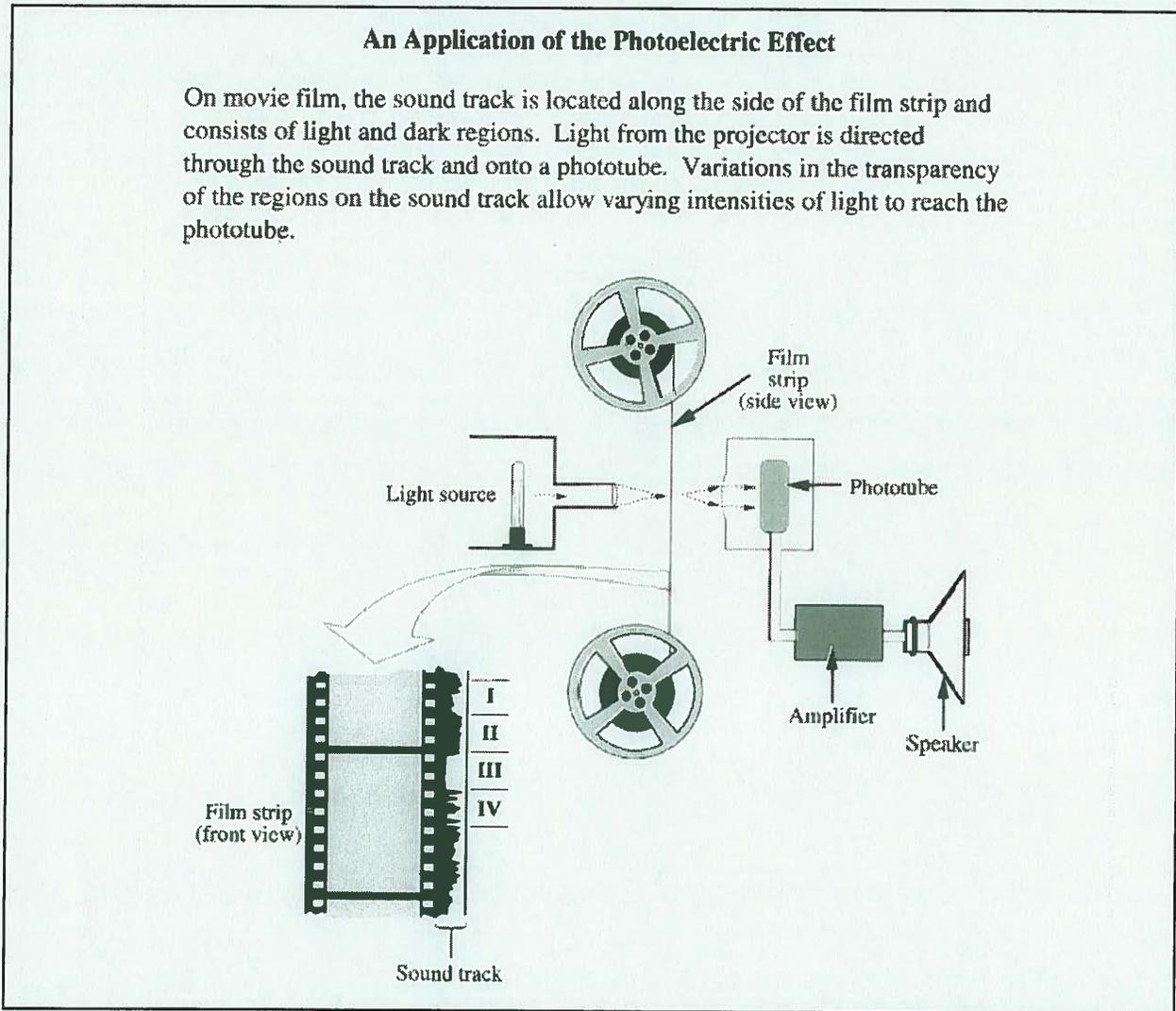
$$E_{K_i} + E_p \rightarrow E_{K_f}$$

$$6.7448275862 \times 10^{-20} + q\Delta V \rightarrow \frac{1}{2}mv^2$$

$$6.74482758... \times 10^{-20} + (1.60 \times 10^{-19})(0.2 \times 10^3) = \frac{1}{2}(9.11 \times 10^{-31})v^2$$

$$v_f = 8.390503 \times 10^6 \text{ m/s}$$

Use the following information to answer Q734:



**Q734:** The region of the sound track that will allow the most electrical current to be produced in the phototube is labelled

- a. I
- b. II
- c. III
- d. IV

More intensity (number of photons) is Region III.  
Each photon can generate one photoelectron.